Vivekananda College of Engineering & Technology, Puttur [A Unit of Vivekananda Vidyavardhaka Sangha Puttur ♥]

Affiliated to VTU, Belagavi & Approved by ALCTE New Delhi

CRMOB

Rev 1.10

ME

25/08/2022

CONTINUOUS INTERNAL EVALUATION 3

Dept ME Sem | Div 4th A Sub Applied Thermodynamics | S Code 18ME42 |
Date | 02/09/2022 | Time | 9/30-11/00 am | Max Marks | 50 | Elective N

Note: Answer any 2 full questions, choosing one full question from each part

QN	Questions	Marks	RET	COs
Magazinian su	PARTA	The second second		
STORES LAND	Explain the working principle of Vapor compression refrigerator with suitable TS, PH & Bock diagram.	7	1.2	CO4
b	A refrigeration system operates on the reversed Carnot cycle. The higher	8	L3	CO4
	temperature of the refrigerant in the system is 50°C and lower			
	temperature is -10°C. The Capacity is to be 10 Tones. Neglect all the losses.			
e	A Simple NH ₃ vapor compression system has a condenser temperature of 30°C & an Evaporator temperature of -15°C. The liquid is sub cooled by		L3	CO4
	10°C. The volumetric efficiency of the compressor is 0.8. Find the			
	following for Dry Compression sub cooled condition: The refrigerating			
L	effect, Flow rate of refrigerant, Volume capacity of compressor, COP & The power per TOR.			
The transfer	OR	Societario esperante de la constitución de la const	distance received	- Abanasian
1	Explain the working principle of Vapor absorption refrigerator with suitable Block diagram		L2	CO4
b	An ideal air refrigeration cycle was designed with following	8	L3	CO4
	specifications with respect to air. Compressor inlet pressure is			
	103.42kPa, Compressor Outlet pressure is 413.7kPa, Temperature at			
	Compressor inlet is -7°C & Temperature at Turbine inlet is 27°C. Find:			
	COP of the cycle, Power required to produce 1 TOR & Air circulation			
	rate if compression and expansion is adiabatic.			
С	An ammonia ice plant operates between a Condenser temperature of	8	L3	CO4
	35°C and an Evaporator temperature of -15°C. It produces 10 tons of ice			
To the state of th	from water at 30°C to ice at -5°C in one day. Find the following for VCR			
	plant: Capacity of the Plant, Mass flow rate of the ammonia. Compressor			
	Power, COP of the Plant, if Cpw= 4.187kJ/kg.K. Cpice = 1.94kJ/kg.K			
	Latent heat of fusion = 335kJ/kg.	7		
on and the second	PART B	-		- Carlos Parado and Carlos Car
a	Derive an expression for Work done of a Reciprocating compresso	r 7	L2	CO4
- 1	without considering Clearance volume for a) adiabatic process b	1	Rodelin	
ļ.	sothermal process			
b	An air compressor takes in air at 1 bar and 20°C and compresses th	e 8	TL3	CO
5	same according to the law pV ^{1.2} = constant. It is then delivered to	0	Auto	
I	eciever at a constant pressure of 10bar. If R = 0.287kI/kg K determine	a		
t	he temperature at the end of compression, Net work consumed and her			
1	ransferred during compression per kg of air supplied.			
C /	A single stage air compressor of 18cm diameter and 120cm stroke lengt	h 8	L	3 CO
r	runs at 120rpm and operates between 1 bar and 10 bar. The lower	et o	Lo	CO
ŧ	emperature of the cycle is 15°C. Compression index is 1.3. Estimate the			
F	Power & final temperature of air for a)Single acting & b) Double acting	iC .		
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4	-	compressor for Polytrophic process.	7	L2	CO4	
		A double acting single stage compressor with piston displacement of 0.04m^3 per stroke operates at 480rpm. The clearance is 4% and it receives air at 100kPa and discharges at 500kPa. The compression and expansion is polytrophic, pV ^{1.3} = Constant. Determine the Power required and the air discharged in m ³ /sec.	8	L3	CO4	
	С	The LP cylinder of a compound air compressor draws in 0.1m ³ of air at a temperature of 15°C and pressure 1 bar. It compresses the air adiabatically to 2 bar and then delivers it in to a receiver where the air is cooled to 25°C. This air is drawn in to a HP cylinder and compressed adiabatically to 5bar and delivered to the receiver. Find the power required when the compressor makes 100rpm. What pressure in the receiver would give the best efficiency assuming other data as above.	8	L3	CO4	

Prepared by:

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